## Slides

Condensed Matter Physics Lecture 19

Green = Filled Shell Atoms L=S=0.
Red = half-filled minus one such that $\mathrm{L}=\mathrm{S}$ and $\mathrm{J}=\mathrm{L}-\mathrm{S}=0$.


| *Lanthanide series |  | Ce | Pr | Nd | $\mathrm{Pm}^{61}$ | $5$ | $\begin{aligned} & \text { and } \\ & \text { Eun } \end{aligned}$ | $\begin{aligned} & \substack{\text { anden } \\ \text { Gd }} \end{aligned}$ | $\begin{aligned} & \substack{\text { and } \\ \text { Tb }} \end{aligned}$ | $\begin{gathered} \text { ack } \\ \text { Dy } \end{gathered}$ | $\begin{aligned} & \text { an mivicio } \\ & \text { Ho } \end{aligned}$ | $\begin{aligned} & { }^{606 m} \\ & \text { Er } \end{aligned}$ | Tm | $\mathrm{Yb}^{70}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| **Actinide series |  |  | $\begin{aligned} & \text { Pat } \\ & \text { and } \end{aligned}$ | $\underset{U}{40}$ | $\begin{aligned} & \text { nen } \\ & \mathrm{No} \end{aligned}$ | $\begin{aligned} & \mathrm{Pa4} \\ & \mathrm{Pu} \end{aligned}$ | $A$ |  | $\begin{gathered} \\ \mathbf{c}^{9720} \end{gathered}$ | ${ }_{\mathrm{Cb}}^{90}$ |  |  | Md | $\stackrel{102}{102}$ |

List of $\mathrm{J}=0$ atoms.

